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Several samples of solid sodium (a good free-electron metal) with different levels of impurities.

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In a crystalline solid, the atoms, ions and molecules are arranged in a Definite repeating pattern, but some defects may occur in the pattern. derivations from perfect arrangement may occur due to rapid cooling or presence of additional particles.

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atoms are positioned in an irregular manor and the solids lack long range order. Many important solid materials are amorphous such as synthetic fibers, plastics, and glasses, but pure solid substances such as elemental phosphorus or sulfur may also exist in amorphous forms.

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